Parasite glycoconjugates. Part 3.¹ Synthesis of substrate analogues of early intermediates in the biosynthetic pathway of glycosylphosphatidylinositol membrane anchors

Sylvain Cottaz,^a John S. Brimacombe *,^a and Michael A. J. Ferguson^b

^a Department of Chemistry, University of Dundee, Dundee DD1 4HN, UK

^b Department of Biochemistry, University of Dundee, Dundee DD1 4HN, UK

Substrate analogues of sodium 1D-6-O-(2-[³H₃]acetamido-2-deoxy- α -D-glucopyranosyl)-myo-inositol 1-[sn-2,3-bis(palmitoyloxy)propyl phosphate] 3, including the lipid-depleted compounds 9, 11 and 15, have been prepared for biological evaluation with a partially purified de-N-acetylase from the bloodstream form of the parasitic protozoan *Trypanosoma brucei*. Such analogues as 2-deoxy- α -D-Glcp-PI 25 and α -D-Glcp-PI 33 have also been prepared and tested as potential inhibitors of this de-N-acetylase.

Introduction

Glycosyl phosphatidylinositol (GPI) membrane anchors are distributed widely among the eukaryotes.² Their principal function is to anchor proteins or oligosaccharides to the outer leaflet of the plasma membrane.^{3.4} Protein-linked GPI anchors and GPI-related glycolipids, such as the lipophosphoglycans and glycoinositol phospholipids of the Leishmania, are particularly abundant in the protozoa.⁵ The tsetse flytransmitted African trypanosomes, which cause human sleeping sickness and a variety of livestock diseases, are able to survive in the host's bloodstream by virtue of a dense cellsurface coat consisting of 10 million copies of a GPI-anchored glycoprotein called the variant surface glycoprotein (VSG).⁶ A schematic representation of the GPI-membrane anchor of the VSG of Trypanosoma brucei, an African protozoan parasite that causes a disease related to sleeping sickness in domestic cattle, is shown in formula 1.4

The complete GPI biosynthetic pathway in trypanosomes has been elucidated, ⁷⁻¹⁴ while that in other protozoa¹⁵ and in mammalian cells¹⁶ appears to be broadly similar. The first step in the biosynthetic pathway involves the transfer of α -D-GlcpNAc from UDP-D-GlcpNAc (where UDP = uridine

5'-diphospho) to form α -D-GlcpNAc-(1 \rightarrow 6)-PI, which is rapidly de-N-acetylated to form α -D-GlcpNH₂-(1 \rightarrow 6)-PI.⁷ Each of the three α -D-Manp residues is then transferred in turn from dolichol phosphate D-mannose to form α -D-Manp-(1 \rightarrow 2)- α -D-Manp-(1 \rightarrow 6)- α -D-Manp-(1 \rightarrow 4)- α -D-GlcpNH₂-(1 \rightarrow 6)-PI,^{8.9} to which is added ethanolamine phosphate (from phosphatidylethanolamine) at the terminal a-D-Manp residue.¹⁰ The resulting structure undergoes a complex series of fattyacid-remodelling reactions¹¹ (to yield an sn-dimyristoyl-PI moiety) before the preassembled GPI precursor (known as glycolipid A) is transferred en bloc to newly synthesized protein.¹² Some a-D-galactosylation of the GPI anchor occurs in the endoplasmic reticulum,13 the principal site of biosynthesis, but mainly in the Golgi apparatus during transport to the surface membrane, ¹⁴ when as many as five α -D-Galp residues may be added to the GPI anchors of T. brucei VSG.² The extraordinary dependence of parasitic protozoa on GPI-anchored proteins and/or GPI-related molecules for survival and infectivity makes the GPI biosynthetic pathway an attractive target for the development of chemotherapeutic agents. This approach is predicated on the belief that disruption of GPI biosynthesis would seriously impair the parasite's ability to survive in the host's bloodstream.







6 R = OBn



Results and discussion

In our initial investigation¹⁷ along these lines, the glycosylphosphatidylinositol 2 was synthesized and, after N-[3H]acetylation (---- 3), was used ¹⁸ as a substrate in the partial purification of the α-D-GlcpNAc-PI de-N-acetylase of the bloodstream form of T. brucei. Having achieved a partial purification of the T. brucei de-N-acetylase, it should now be possible to probe its substrate specificity by using synthetic substrate analogues, and also to examine potential inhibitors of the enzymic de-N-acetylation of the close substrate analogue 3, which differs from the enzyme's natural substrate only in that the lipid component is esterified with palmitic acid rather than other long-chain fatty acids. Routes to the substrate analogues 9, 11 and 15 and the potential inhibitors 25 and 33 are described herein. The substrate analogues 9, 11 and 15 were chosen with a view to assessing the roles of the lipid, phospholipid and fatty acid moieties, respectively, in substrate recognition, whereas 2deoxy- α -D-Glcp-PI 25 and α -D-Glcp-PI 33 retain all the components of the substrate analogue 3 except that the vital NHAc group at C-2 of the sugar moiety is replaced by either H or OH.

Although several routes have now been devised ^{19.20} for the synthesis of 1D-6-O-(2-amino-2-deoxy- α -D-glucopyranosyl)myo-inositol 1-phosphate **8**, which is also of interest for its possible insulin-like activity,²⁰ we chose to use an adaptation of the hydrogen phosphonate procedure.²¹ Thus, the glycosyl



inositol derivative 417 reacted with phosphorous acid in pyridine to give, after conversion into the triethylammonium (TEA) salt, the hydrogen phosphonate 5. Conventional treatment of the latter compound with benzyl alcohol in the presence of pivaloyl chloride in pyridine^{22.23} furnished the phosphonic diester 6, as a mixture of diastereoisomers. Oxidation of the phosphonic diester 6 in situ with iodine in aq. pyridine²¹ gave the phosphoric diester, which was isolated and characterised as the TEA salt 7. Hydrogenolysis of the freeacid form of the phosphoric diester 7 in the presence of 20% palladium hydroxide on carbon yielded the deprotected compound 8, whose conversion into the N-tritritioacetyl derivative 9 for enzymic studies has been described elsewhere.¹⁸ Also prepared from the glycosyl-inositol derivative 4^{17} by similar hydrogenolysis was the dephosphorylated material 10. The preparation of the N-tritritioacetyl derivative 11 from compound 10 and enzymic studies thereon have been described elsewhere.18

Deacylation of the previously described ¹⁷ di-O-palmitoyl compound **12** with sodium methoxide gave the glycerophosphate derivative **13**, which on catalytic hydrogenolysis produced the phosphoric diester **14** required for conversion ¹⁸ into the *N*-tritritioacetyl derivative **15**. Biological assay of this and the other tritiated compounds **3**, **9** and **11** with the *T. brucei* de-N-acetylase was based on the initial rate of release of $[^{3}H]$ acetic acid.¹⁸

The first step in the synthesis of the 2'-deoxy analogue 25 of the proven¹⁸ substrate 3 for the *T. brucei* de-N-acetylase involved the coupling of 3,4,6-tri-*O*-benzyl-D-glucal 16 with the 1D-myo-inositol derivative 17^{17} in the presence of triphenylphosphine-hydrogen bromide.²⁴ The α -coupled product 18 $(J_{1',2'ax} 3.5, J_{1',2'eq} \sim 1 \text{ Hz})$, obtained in 56% yield, was then treated with ammonium cerium(iv) nitrate ²⁵ (CAN) in aq. acetonitrile to remove the 4-methoxybenzyl group at O-1 of the D-myo-inositol residue (\longrightarrow 19). Thereafter the hydrogen phosphonate approach²¹ was used to introduce the phospholipid component at O-1. This involved the reaction of compound 19 with the hydrogen phosphonate 20²² in the presence of pivaloyl chloride in pyridine^{22.23} to give the





phosphonic diester 21 as a mixture of diastereoisomers. Oxidation of this diester with iodine in aq. pyridine²¹ afforded the phosphoric diester 22, which was isolated and characterised at the TEA salt 23. Hydrogenolysis of the corresponding sodium salt 24 in the presence of palladium hydroxide on carbon provided the fully deprotected 2-deoxy- α -D-Glcp-PI 25 (isolated as the TEA salt).

For the synthesis of α -D-Glcp-PI 33, the 1D-myo-inositol derivative 17^{17} was coupled in diethyl ether with the trichloroacetimidate 26^{26} in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) to give the fully protected α -D-glucopyranosylinositol 27 ($J_{1'.2'}$ 3.4 Hz) in 76% yield. Thereafter, the compound 27 was taken through an identical sequence of reactions, viz. 27 \longrightarrow 28 \longrightarrow 29 \longrightarrow 30 \longrightarrow 31 \longrightarrow 32 \longrightarrow 33, to that used in preparing 2-deoxy- α -Glcp-PI 25.

The results of biological evaluation of α -D-GlcNAcp-PI 3 and the substrate analogues 9, 11 and 15 have been reported elsewhere.¹⁸ Briefly, α -D-GlcpNAc-PI 3 proved to be a good exogenous substrate for a partially purified *T. brucei* de-N-



acetylase. The analogue 15, lacking the fatty acid components, was a relatively poor substrate for the enzyme, indicating that the lipid component plays an important role in substrate recognition and/or presentation to the enzyme in detergent micelles. This view is reinforced by the fact that neither of the substrate analogues 9 or 11, which lack the lipid and phospholipid components, respectively, was recognised by the enzyme. α -D-GlcpNH₂-PI 2,¹⁷ a product of enzymic action, showed weak inhibition, implying that product inhibition is unlikely to play a regulatory role in the GPI biosynthetic pathway. The substrate/product analogues 2-deoxy- α -D-Glcp-PI 33 were, respectively, much better and much poorer inhibitors of the hydrolysis of α -D-GlcpNH₂-PI 2.

Recent studies 27 on the first α -D-mannosyl transferase of the GPI biosynthetic pathway have demonstrated that, in comparison with the close substrate analogue 2, the delipidated compounds 8 and 10 were poor substrates. The compound 14, which differs from the natural substrate in having the fatty acid residues removed, was a slightly better substrate than was either of the compounds 8 or 10, whereas neither 2-deoxy- α -D-Glcp-PI 25 nor α -D-Glcp-PI 33 was recognised by the enzyme. Details of these studies, which highlight the importance of the NH₂ group in substrate recognition by this α -D-mannosyl transferase, will be published elsewhere in due course.

Experimental

Mps were determined on a Reichert hot-plate apparatus and are uncorrected. TLC was performed on silica gel 60 GF_{254}

(Merck) and spots were detected with UV light or by charring with dil. sulfuric acid as appropriate. Flash-column chromatography was performed on silica gel 60 (230–400 mesh, Merck). ¹H NMR spectra were recorded on Bruker AM 200 MHz or AC 500 MHz spectrometers with deuteriochloroform as solvent and tetramethylsilane as internal reference, unless otherwise indicated; *J* values are given in Hz. Mass spectra were measured with VG 250/70SE or VG Fisons Quattro instruments. Optical rotations were obtained using a Perkin-Elmer 141 polarimeter at ambient temperature, and are given in units of 10^{-1} deg cm² g⁻¹.

Triethylammonium 1D-6-*O*-(2-azido-3,4,6-tri-*O*-benzyl-2deoxy-α-D-glucopyranosyl)-2,3,4,5-tetra-*O*-benzyl-*myo*-inositol 1-(hydrogenphosphonate) 5

A solution of compound 4¹⁷ (85 mg, 85 µmol; previously dried by co-evaporation with pyridine) and pivaloyl chloride (0.06 cm³, 0.49 mmol) in 2 mol dm⁻³ phosphorous acid in pyridine (0.45 cm³, 0.9 mmol) was stirred at room temp. for 3 h before the reaction was stopped by the addition of 1 mol dm⁻³ aq. TEA hydrogen carbonate. The resulting solution was diluted with diethyl ether, and the ethereal solution was washed twice with 1 mol dm⁻³ aq. TEA hydrogen carbonate, dried (MgSO₄), and concentrated under reduced pressure. Flash-column chromatography [chloroform-methanol (7:3)] of the residue gave the TEA hydrogen phosphonate derivative 5 (75 mg, 76%), $[\alpha]_{\rm D}$ + 63 (c 1.4, CHCl₃); $\delta_{\rm H}$ 1.22 (9 H, t, 3 × CH₂Me), 2.93 (6 H, q, $3 \times CH_2$ Me), 3.15 (1 H, dd, $J_{5'.6'a}$ 1.9, $J_{6'a.6'b}$ 11.7, 6'-H^a), 3.26 (1 H, dd, $J_{5'.6'b}$ 3.9, 6'-H^b), 3.27 (1 H, dd, $J_{1'.2'}$ 3.9, $J_{2'.3'}$ 10.2, 2'-H), 3.45 (1 H, t, $J_{4.5} = J_{5.6} = 9.3$, 5-H), 3.57 (1 H, dd, $J_{2.3}$ 2.5, $J_{3,4}$ 10.0, 3-H), 3.70 (1 H, t, $J_{3',4'} = J_{4',5'} = 9.3, 4'$ -H), 3.98 (1 H, t, 3'-H), 4.06 (1 H, m, 5'-H), 4.10 (1 H, t, 4-H), 4.20 (1 H, dd, J_{1.6} 9.2, 1-H), 4.29 (1 H, t, 6-H), 4.63 (1 H, t, 2-H), 5.74 (1 H, d, 1'-H), 7.13 (1 H, d, J_{H.P} 622, HP) and 7.00-7.49 (35 H, m, $7 \times Ph$).

Triethylammonium 1D-6-*O*-(2-azido-3,4,6-tri-*O*-benzyl-2deoxy-α-D-glucopyranosyl)-2,3,4,5-tetra-*O*-benzyl-*myo*-inositol 1-(benzyl phosphate) 7

A mixture of the hydrogenphosphonate 5 (13 mg, 11 µmol), benzyl alcohol (5 mm³, 48 µmol) and pivaloyl chloride (8 mm³, 65 μ mol) in dry pyridine (0.25 cm³) was stirred at room temp. for 2 h to give the diastereoisomeric hydrogenphosphonates 6. Oxidation of the hydrogenphosphonates 6 was accomplished by the addition of 0.1 mol dm^{-3} iodine in 2% aq. pyridine (0.2 cm³) to the mixture, which was then stirred at room temp. for 1 h and diluted with diethyl ether. The resulting solution was washed with 5% aq. sodium hydrogen sulfite, dried (MgSO₄), and concentrated under reduced pressure. After flash-column chromatography [first with chloroform and then with chloroform-methanol (9:1)], an ethereal solution of the residue was washed twice with 1 mol dm⁻³ aq. TEA hydrogen carbonate, dried (MgSO₄), and concentrated under reduced pressure to give the TEA phosphate derivative 7 (12 mg, 85%), $[\alpha]_{\rm D}$ +45 (c 1.2, CHCl₃); $\delta_{\rm H}$ 1.17 (9 H, t, 3 × CH₂Me), 2.87 (6 H, q, 3 × CH_2 Me), 3.19 (1 H, dd $J_{1',2'}$ 3.4, $J_{2',3'}$ 10.2, 2'-H), $3.39 (2 \text{ H}, \text{m}, 6'-\text{H}_2), 3.46 (1 \text{ H}, \text{t}, J_{4.5} = J_{5.6} = 9.5, 5-\text{H}), 3.52$ $(1 \text{ H}, \text{dd}, J_{3,4}, 9.5, 3-\text{H}), 3.71 (1 \text{ H}, t, J_{3',4'} = J_{4',5'} = 9.5, 4'-\text{H}),$ 4.02 (1 H, t, 3'-H), 4.08 (1 H, t, 4-H), 4.14 (1 H, m, 5'-H), 4.35 (2 H, m, 1-, 6-H), 4.74 (1 H, t, 2-H), 5.03 (2 H, m, POCH₂Ph), 5.92 (1 H, d, 1'-H) and 6.97–7.40 (40 H, m, 8 × Ph); δ_{P} (CDCl₃; ext. ref. 85% H₃PO₄) -0.39 (with ¹H heteronuclear decoupling).

1D-6-*O*-(2-Amino-2-deoxy-α-D-glucopyranosyl)-*myo*-inositol 1-dihydrogen phosphate) 8

A solution of the free-acid form of the phosphoric diester 7 (prepared from the TEA salt, 12 mg, 9.4 μ mol) in chloroform-

methanol (2 cm³; 1:3) containing 20% palladium hydroxide on carbon (10 mg) was shaken under a slight overpressure of hydrogen at room temp. for 20 h and was then percolated through a short column packed with a layer of Celite on top of silica gel, with washing with methanol. The filtrate and the washings were combined, and concentrated under reduced pressure, and the residue was subjected to flash-column chromatography [acetonitrile-water (7:3) and then (6.5:3.5)] to give the phosphate **8** (3 mg, 75%), $[\alpha]_D + 75$ (c 0.2, H₂O), (lit.,¹⁹ + 73 (c 0.6, H₂O); ES⁻ MS m/z 420 (M⁻ - H).

1D-6-O-(2-Amino-2-deoxy-\alpha-D-glucopyranosyl)-myo-inositol 10 A solution of compound 4¹⁷ (15 mg, 15 µmol) in chloroformmethanol (3 cm³; 1:2) containing 20% palladium hydroxide on carbon (15 mg) was shaken under a slight overpressure of hydrogen at room temperature for 12 h and was then filtered through a Celite pad and concentrated under reduced pressure. The residue was dissolved in water, and the aqueous solution was filtered and freeze-dried to give the glycosyl inositol 10 (5 mg, 97.5%), [α]_D +82 (*c* 0.5, MeOH); $\delta_{\rm H}$ (D₂O; int. ref. Me₂CO at $\delta_{\rm H}$ 2.07) 3.19 (1 H, dd, $J_{1'.2'}$ 3.4, $J_{2'.3'}$ 10.7, 2'-H), 3.23 (1 H, m, 5-H), 3.37 (1 H, dd, $J_{2.3}$ 3.0, $J_{3.4}$ 9.8, 3-H), 3.40 (1 H, t, $J_{3'.4'} = J_{4'.5'} = 9.4$, 4'-H), 3.49 (1 H, t, $J_{3.4} = J_{4.5} = 9.8$, 4-H), 3.59 (2 H, m, 1-, 6-H), 3.67 (2 H, m, 6'-H₂), 3.76 (1 H, dd, 3'-H), 3.87 (1 H, dd, $J_{1.2}$ 2.1, 2-H), 3.92 (1 H, m, 5'-H) and 5.29 (1 H, d, 1'-H). FAB⁺ MS *m*/z 342 (MH⁺).

1D-6-O-(2-Amino-2-deoxy-α-D-glucopyranosyl)-*myo*-inositol 1-(*sn*-2,3-dihydroxypropyl hydrogen phosphate) 14

A solution of compound 12 (as the sodium salt,¹⁷ 45 mg, 27.3 μ mol) and 1 mol dm⁻³ sodium methoxide in methanol (30 mm³) in diethyl ether-methanol (3 cm³; 1:2) was stirred at room temp. for 30 min and was then deionised with Amberlite 1R 120 (H⁺) resin. Removal of the solvents under reduced pressure and flash chromatography [first with chloroform and then with chloroform-methanol (9:1)] of the residue gave the deacylated compound 13 (30 mg, 95.5%); FAB⁺ MS *m/z* 1151 (M⁺ - H).

A solution of compound 13 (30 mg, 26 µmol) in methanol (2 cm³) containing 20% palladium hydroxide on carbon (30 mg) was shaken under a slight overpressure of hydrogen at room temp. for 12 h and was then processed as described in the previous experiment. An aqueous solution of the residue was passed down a short column of Sephadex LH20-100 and the eluent was freeze-dried to give the phosphoric diester 14 (9 mg, 67%), $[\alpha]_D$ + 54 (c 0.9, H₂O); $\delta_H(D_2O)$; int. ref. Me₂CO at δ 2.07) 3.20 (1 H, dd, $J_{1'.2'}$ 3.8, $J_{2'.3'}$ 10.4, 2'-H), 3.27 (1 H, t, $J_{4.5} = J_{5.6} = 9.3, 5$ -H), 3.39 (1 H, t, $J_{3'.4'} = J_{4'.5'} = 9.9, 4'$ -H), 3.40 (1 H, dd, J_{2.3} 2.3, J_{3.4} 10.4, 3-H), 3.45 and 3.52 (2 H, m, 3-H₂ propyl), 3.54 (1 H, dd, 4-H), 3.67 (2 H, m, 6'-H₂), 3.75 (1 H, dd, 3'-H), 3.77 (1 H, m, 2-H propyl), 3.79 (1 H, t, $J_{1.6} =$ $J_{5.6} = 9.9, 6$ -H), 3.84 (2 H, m, 1-H₂ propyl), 3.96 (1 H, m, 5'-H), 4.04 (1 H, t, $J_{1.2} = J_{2.3} = 2.3, 2$ -H), 4.08 (1 H, m, 1-H) and 5.40 (1 H, d, 1'-H); $\delta_{P}(D_{2}O; \text{ ext. ref. 85\% H}_{3}PO_{4})$ 4.25 (with ¹H heteronuclear decoupling). ES⁻ MS m/z 494 (M⁻ – H).

2,3,4,5-Tetra-O-benzyl-1D-6-O-(3,4,6-tri-O-benzyl-2-deoxy-α-D-arabino-hexopyranosyl)-myo-inositol 19

A mixture of 3,4,6-tri-O-benzyl-D-glucal 16 (Fluka, 40 mg, 96 µmol), the D-myo-inositol derivative 17¹⁷ (60 mg, 91 µmol), and triphenylphosphine-hydrogen bromide (1 mg, 2.9 µmol) in dry dichloromethane (0.35 cm³) was stirred at room temp. for 24 h. Flash-column chromatography [cyclohexane-diethyl ether (3:1)] of the reaction mixture gave the α -coupled derivative 18 (55 mg, 56%), [α]_D + 38 (c 5.5, CHCl₃); δ _H(*inter alia*) 1.63 (1 H, ddd, $J_{1'.2'ax}$ 3.5, $J_{2'ax.3'}$ 11.9, $J_{2'ax.2'eq}$ 12.6, 2'-H^{ax}), 2.19 (1 H ddd, $J_{1'.2'eq} \sim 1$, $J_{2'eq.3'}$ 4.9, 2'-H^{eq}), 3.78 (3 H, s, OMe), 5.59 (1 H, dd, 1'-H) and 6.83–7.40 (39 H, C₆H₄ and 7 × Ph).

A solution of compound 18 (30 mg, 28 µmol) in acetonitrile-

water (1 cm³; 9:1) containing CAN (80 mg, 144 µmol) was stirred at 0 °C for 1 h and was then diluted with diethyl ether. The ethereal solution was washed twice with aq. sodium chloride, dried (MgSO₄), and concentrated under reduced pressure. Flash-column chromatography [toluene-butanone (9:1)] of the residue gave the 2-deoxyglycosylinositol **19** (18 mg, 67.5%), mp 108–110 °C (from diethyl ether-hexane); $[\alpha]_D + 71$ (*c* 1.8, CHCl₃) (Found: C, 76.6; H, 6.85. C₆₁H₆₄O₁₀ requires C, 76.5; H, 6.7%); $\delta_{\rm H}(inter alia)$ 1.69 (1 H, ddd, $J_{1'.2'ax}$ 3.5, $J_{2'ax,3'}$ 11.5, $J_{2'ax,3'eq}$ 12.7, 2'-H^{ax}), 2.33 (1 H, ddd, $J_{1'.2'eq} \sim 1$, $J_{2'eq,3'}$ 4.9, 2'-H^{eq}), 5.64 (1 H, dd, 1'-H) and 6.99–7.36 (35 H, m, 7 × Ph).

Triethylammonium 1D-2,3,4,5-tetra-O-benzyl-6-O-(3,4,6-tri-Obenzyl-2-deoxy-α-D-*arabino*-hexopyranosyl)-*myo*-inositol 1-[*sn*-2,3-bis(palmitoyloxy)propyl phosphate] 23

A mixture of compound 19 (66 mg, 69 µmol), sn-2,3bis(palmitoyloxy)propyl hydrogen hydrogenphosphonate²² 20 (50 mg, 79 µmol) and pivaloyl chloride (40 mm³, 0.32 mmol) in anhydrous pyridine (3 cm³) was stirred at room temp. for 10 min to form a mixture of the diastereoisomeric hydrogenphosphonates 21. Oxidation of the hydrogenphosphonates was accomplished by the addition of 0.1 mol dm^{-3} iodine in 2% aq. pyridine (1.6 cm³) to the mixture, which was then stirred at room temp. for 10 min and diluted with diethyl ether. The ethereal solution was washed with 5% aq. sodium hydrogen sulfite, dried (MgSO₄), and concentrated under reduced pressure. Flash-column chromatography [first with chloroform and then with chloroform-methanol (19:1)] of the residue gave the phosphoric diester 22, which was taken up in diethyl ether, and the ethereal solution was washed with 1 mol dm⁻³ aq. TEA hydrogen carbonate, and concentrated under reduced pressure to give the TEA phosphate derivative 23 (94 mg, 81%), $[\alpha]_D$ +40 (c 1.3, CHCl₃); $\delta_{\rm H}$ 0.88 (6 H, t, J 7.3, 2 × CH₂Me), 1.19 (9 H, t, $3 \times CH_2Me$), 1.24 (48 H, m, $2 \times [CH_2]_{12}$), 1.56 (4 H, m, 10.0, 3-H), 3.62 (1 H, t, $J_{3',4'} = J_{4',5'} = 9.0, 4'$ -H), 3.97 (1 H, dt, 5'-H), 4.02 (1 H, m, 3'-H), 4.07 (1 H, t, 4-H), 4.10 and 4.37 (4 H, 2 m, 2 × CH₂ propyl), 4.19 (1 H, m, 1-H), 4.26 (1 H, dd, $J_{1.6}$ 7.8, 6-H), 4.58 (1 H, br t, 2-H), 5.24 (1 H, m, 2-H propyl), 5.64 (1 H, dd, $J_{1',2'ax}$ 2.5, 1'-H) and 6.99–7.45 (35 H, m, 7 × Ph); $\delta_{\rm P}({\rm CDCl}_3;$ ext. ref. 85% H₃PO₄) 0.08 (with ¹H heteronuclear decoupling).

Triethylammonium 1D-6-O-(2-deoxy-α-D-arabino-hexopyranosyl)-myo-inositol 1-[sn-2,3-bis(palmitoyloxy)propyl phosphate] 25

The sodium salt 24 was obtained quantitatively upon stirring of a solution of the TEA phosphate 23 in methanol with Amberlite DPI (Na⁺) resin for 3 h, filtration, and concentration of the filtrate under reduced pressure. A solution of the sodium salt 24 (47 mg, 29 µmol) in methanol-tetrahydrofuran (THF) (4 cm³; 1:1) containing 20% palladium on carbon (50 mg) was shaken under a slight overpressure of hydrogen at room temp. for 20 h and was then percolated through a short column packed with a layer of Celite on top of silica gel, with washing with methanol-THF (1:1). The eluent and washings were combined, and concentrated under reduced pressure. Flash-column chromatography [first with chloroform and then with chloroformmethanol (1:1)] gave the debenzylated compound, which was taken up in butan-1-ol and shaken with 1 mol dm⁻³ aq. TEA hydrogen carbonate. Concentration of the organic phase under reduced pressure gave the TEA phosphate derivative 25 (17 mg, 55%), $[\alpha]_D$ +28 [c 0.8, CHCl₃-MeOH (9:1)]; δ_H (inter alia) $(CDCl_3-CD_3OD, 9:1), 0.88(6H, t, J7.3, 2 \times CH_2Me), 1.28(48)$

H, m, 2 × $[CH_2]_{12}$, 1.32 (9 H, t, 3 × CH_2Me), 1.58 (1 H, ddd, 2'-H^{ax}), 1.60 (4 H, m, COCH₂CH₂), 2.27 (1 H, ddd, $J_{1',2'eq} \sim I$, $J_{2'eq,3'}$ 4.5, 2'-H^{eq}), 2.29 and 2.33 (4 H, 2 t, 2 × COCH₂), 4.15 (1 H, m, 1-H^a propyl), 4.39 (1 H, dd, $J_{1b,2}$ 3.2, $J_{1a,1b}$ 12.2, 1-H^b propyl), 4.23 (1 H, br t, 2-H), 5.21 (1 H, dd, 1'-H) and 5.23 (1 H, m, 2-H propyl); δ_{P} (CDCl₃-CD₃OD, 9:1) -0.15 (with ¹H heteronuclear decoupling). ES⁻ MS m/z 955.5 (M⁻ – TEA).

$1D-2,3,4,5-Tetra-O-benzyl-1-O-(4-methoxybenzyl)-6-O-(2,3,4,6-tetra-O-benzyl-\alpha-D-glucopyranosyl)-myo-inositol\ 27$

To a cooled (-60 °C) mixture of the trichloroacetimidate **26**²⁶ (200 mg, 290 µmol), the D-myo-inositol derivative **17**¹⁷ (150 mg, 230 µmol) and powdered 3 Å molecular sieves (0.5 g) in anhydrous diethyl ether (5 cm³) under nitrogen was added dropwise 0.02 mol dm⁻³ TMSOTf in anhydrous diethyl ether (0.25 cm³). The reaction mixture was stirred at -60 °C for 0.5 h, neutralised with triethylamine (0.5 cm³), and then percolated through a short column of silica gel, with washing with diethyl ether. The combined eluent was concentrated under reduced pressure and the residue was subjected to flash-column chromatography [toluene-butanone (50:1)] to give the D-glucopyranosylinositol **27** (196 mg, 76%), $[\alpha]_D$ +38 (c 1.5, CHCl₃); $\delta_{\rm H}(inter alia)$ 3.75 (3 H, s, OMe) and 5.85 (1 H, d, $J_{1'.2'}$ 3.4, 1'-H); $\delta_{\rm C}({\rm CDCl}_3)$ 96.36 (C-1'). ES⁺ MS *m*/*z* 1205.6 (M⁺ + Na).

1D-2,3,4,5-Tetra-O-benzyl-6-O-(2,3,4,6-tetra-O-benzyl-α-Dglucopyranosyl)-myo-inositol 28

A solution of compound **27** (73 mg, 62 µmol) in 1% trifluoroacetic acid in anhydrous dichloromethane (8 cm³) was stirred at room temp. for 1.5 h, and was then neutralised with triethylamine (0.5 cm³) and concentrated under reduced pressure. Flash-column chromatography [cyclohexane-diethyl ether (3:1)] of the residue gave the alcohol **28** (53 mg, 81%), $[\alpha]_D$ +45 (c 2, CHCl₃); δ_H 3.09 (1 H, dd, $J_{5'.6'a}$ 1.7, $J_{6'a.6'b}$ 11.0, 6'-H^a), 3.38 (1 H, t, $J_{4.5} = J_{5.6} = 9.0, 5$ -H), 3.40 (1 H, dd, $J_{5'.6'b}$ 2.6, 6'-H^b), 3.45 (1 H, dd, $J_{2.3}$ 2.5, $J_{3.4}$ 10.2, 3-H), 3.59 (1 H, dd, $J_{1'2'}$ 3.5, $J_{2'.3'}$ 9.6, 2'-H), 3.60 (1 H, dd, $J_{1.2}$ 2.5, 1-H), 3.73 (1 H, t, $J_{3'.4'} = J_{4'.5'} = 9.3, 4'$ -H), 3.90 (1 H, dt, 5'-H), 3.97 (1 H, t, 6-H), 4.01 (1 H, t, 3'-H), 4.05 (1 H, t, 2-H), 4.10 (1 H, t, 4-H), 4.05-4.96 (12 H, m, 8 × CH₂Ph), 5.17 (1 H, d, 1'-H) and 6.99-7.45 (40 H, m, 8 × Ph). ES⁺ MS *m*/*z* 1085.5 (M⁺ + Na).

Triethylammonium 1D-2,3,4,5-tetra-O-benzyl-6-O-(2,3,4,6-tetra-O-benzyl-α-D-glucopyranosyl)-*myo*-inositol 1-[*sn*-2,3-bis(palmitoyloxy)propyl phosphate] 31

A mixture of compound **28** (42 mg, 39.5 µmol), the hydrogenphosphonate **20**²² (28 mg, 44 µmol) and pivaloyl chloride (25 mm³, 0.2 mmol) in anhydrous pyridine (2 cm³) was stirred at room temp. for 20 min to form the diastereoisomeric hydrogenphosphonates **29**, which were converted into the phosphoric diester **30** and thereafter into the TEA phosphate derivative **31** (43 mg, 61%), $[\alpha]_D$ +36 (*c* 1.45, CHCl₃), as described for the corresponding 2'-deoxy derivative **21**; δ_H (*inter alia*) 0.88 (6 H, t, J 7.3, 2 × CH₂Me), 1.11 (9 H, t, 3 × CH₂Me), 1.20 (48 H, m, 2 × [CH₂]₁₂), 1.52 (4 H, m, 2 × COCH₂CH₂), 2.17 and 2.20 (4 H, 2 t, 2 × COCH₂), 2.79 (6 H, q, 3 × CH₂Me), 3.44 (2 H, m, 6'-H₂), 3.56 (1 H, dd, 2'-H), 3.61 (1 H, t, 4'-H), 4.01 (1 H, t, 3'-H), 4.35 (1 H, m, 5'-H), 5.20 (1 H, m, 2-H propyl), 6.01 (1 H, d, $J_{1',2'}$ 3.4, 1'-H) and 6.89–7.52 (40 H, m, 8 × Ph). ES⁻ MS *m*/z 1692.1 (M⁻ – TEA).

Sodium 1D-6-O- α -D-glucopyranosyl-*myo*-inositol 1-[*sn*-2,3-bis-(palmitoyloxy)propyl phosphate] 33

A solution of the sodium salt 32 (30 mg, 17.7 μ mol; prepared from the TEA derivative 31 as described for compound 24) in methanol-dichloromethane (3 cm³; 1:1) containing 20%

palladium hydroxide on carbon (10 mg) was shaken under a slight overpressure of hydrogen at room temp. for 12 h, and was then filtered with the aid of Celite and concentrated under reduced pressure to give the a-D-Glcp-PI 33 (10 mg, 96%), [a]_D +12.5 (c 1, Me_2SO), as a solid, which TLC [chloroformmethanol (3:2)] indicated to have a purity >90%; $\delta_{\rm H}$ [(CD₃)₂SO] 0.84 (6 H, t, J 7.3, 2 × CH₂Me), 1.23 (48 H, m, $2 \times [CH_2]_{12}$, 1.49 (4 H, m, 2 × COCH₂CH₂), 2.27 (4 H, m, 2 × COCH₂), 4.10 (1 H, dd, $J_{1a,2}$ 7.5, $J_{1a,1b}$ 11.2, 1-H^a propyl), 4.29 (1 H, dd, $J_{1b,2}$ 2.5, 1-H^b propyl), 4.98 (1 H, d, $J_{1',2'}$ 3.5, 1'-H) and 5.12 (1 H, m, 2-H propyl). ES⁻ MS m/z 971.5 (M⁻ – Na). This material was further purified by semi-preparative HPLC, essentially as described ¹⁸ for α-D-GlcpNAc-PI 3, before being used in inhibition studies¹⁸ with the T. brucei de-Nacetylase.

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References

- 1 Part 2. S. Cottaz, J. S. Brimacombe and M. A. J. Ferguson, Carbohydr. Res., in the press.
- 2 P. T. Englund, Annu. Rev. Biochem., 1993, 62, 121; M. J. McConville and M. A. J. Ferguson, Biochem. J., 1993, 294, 305; D. A. Brown, Curr. Opin. Immunol., 1993, 5, 338.
- 3 M. A. J. Ferguson and A. F. Williams, Annu. Rev. Biochem., 1988, 57, 285; M. A. J. Ferguson, Curr. Opin. Struct. Biol., 1991, 1, 522; Biochem. Soc. Trans., 1992, 20, 243; M. J. McConville, Cell Biol. Int. Rep., 1991, 15, 779.
- 4 M. A. J. Ferguson, S. W. Homans, R. A. Dwek and T. W. Rademacher, Science, 1988, 239, 753; M. A. J. Ferguson, J. S. Brimacombe, S. Cottaz, R. A. Field, L. S. Güther, S. W. Homans, M. J. McConville, A. Mehlert, K. G. Milne, J. E. Ralton, Y. A. Roy, P. Schneider and N. Zitzmann, Parasitology, 1994, 108, S45
- 5 M. A. J. Ferguson, W. J. Masterson, S. W. Homans and M. J. McConville, Cell Biol. Int. Rep., 1991, 15, 991; S. J. Turco and A. Descoteaux, Annu. Rev. Microbiol., 1992, 46, 65.
- 6 G. A. M. Cross, Annu. Rev. Cell Biol., 1990, 6, 1.
- 7 T. L. Doering, W. J. Masterson, P. T. Englund and G. W. Hart, J. Biol. Chem., 1989, 264, 11168.
- 8 W. J. Masterson, T. L. Doering, G. W. Hart and P. T. Englund, Cell, 1989, 56, 793.
- 9 A. K. Menon, S. Mayor and R. T. Schwarz, EMBO J., 1990, 9, 4249;

A. K. Menon, R. T. Schwarz, S. Mayor and G. A. M. Cross, J. Biol. Chem., 1990, 265, 9033.

- 10 A. K. Menon, M. Eppinger, S. Mayor and R. T. Schwarz, EMBO J., 1993, 12, 1907.
- 11 W. J. Masterson, J. Raper, T. L. Doering, G. W. Hart and P. T. Englund, Cell, 1990, 62, 73
- 12 J. D. Bangs, D. Hereld, J. L. Krakow, G. W. Hart and P. T. Englund, Proc. Natl. Acad. Sci. USA, 1985, 82, 3207; M. A. J. Ferguson, M. Duszenko, G. S. Lamont, P. Overath and G. A. M. Cross, J. Biol. Chem., 1986, 261, 356.
- 13 S. Mayor, A. K. Menon and G. A. M. Cross, J. Biol. Chem., 1992, 267, 754.
- 14 J. D. Bangs, T. L. Doering, P. T. Englund and G. W. Hart, J. Biol. Chem., 1988, 263, 17697.
- 15 S. Tomavo, J.-F. Dubremetz and R. T. Schwarz, J. Biol. Chem., 1992, 267, 11721, 21446; P. Gerold, A. Dieckmann-Schuppert and R. T. Schwarz, J. Biol. Chem., 1994, 269, 2597.
- 16 S. Hirose, R. P. Mohney, S. C. Mutka, L. Ravi, D. R. Singleton, G. Perry, A. M. Tartakoff and M. E. Medof, J. Biol. Chem., 1992, 267, 5272; T. Kamitani, A. K. Menon, Y. Hallaq, C. D. Warren and E. T. H. Yeh, *J. Biol. Chem.*, 1992, **267**, 24611; A. Puoti and A. Conzelmann, *J. Biol. Chem.*, 1993, **268**, 7215.
- 17 S. Cottaz, J. S. Brimacombe and M. A. J. Ferguson, J. Chem. Soc., Perkin Trans. 1, 1993, 2945.
- 18 K. G. Milne, R. A. Field, W. J. Masterson, S. Cottaz, J. S. Brimacombe and M. A. J. Ferguson, J. Biol. Chem., 1994, 269, 16403.
- 19 C. Jaramillo, J.-L. Chiara and M. Martín-Lomas, J. Org. Chem., 1994, 59, 3135.
- 20 A. Zapata, Y. León, J. M. Mato, I. Varelo-Nieto, S. Penadés and M. Martín-Lomas, Carbohydr. Res., 1994, 264, 21; S. V. Ley and L. L. Yeung, Synlett., 1992, 997.
- 21 A. V. Nikolaev, I. A. Ivanova, V. N. Shibaev and N. K. Kochetkov, Carbohydr. Res., 1990, 204, 65, and references cited therein.
- 22 I. Lindh and J. Stawiński, *J. Org. Chem.*, 1989, **54**, 1338. 23 T. M. Slaghek, A. A. M. Maas, J. P. Kamerling and J. F. G. Vliegenthart, Carbohydr. Res., 1991, 211, 25; B. C. Froehler and M. D. Matteucci, Tetrahedron Lett., 1986, 27, 467.
- 24 V. Bolitt, C. Mioskowski, S.-G. Lee and J. R. Falck, J. Org. Chem., 1990, 55, 5812; N. Kaila, M. Blumenstein, H. Bielawska and R. W. Franck, J. Org. Chem., 1992, 57, 4576.
- 25 R. Johansson and B. Samuelsson, J. Chem. Soc., Perkin Trans. 1, 1984. 2371.
- 26 R. R. Schmidt, J. Michel and M. Roos, Liebigs Ann. Chem., 1984, 1343.
- 27 T. K. Smith and M. A. J. Ferguson, unpublished results.

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